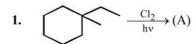
Hydrocarbons



Find the number of monochloro derivatives formed (excluding stereoisomers) in the above reaction.

2. Find the total number of trichloroderivatives of the

compound (excluding stereoisomers).

- **3.** How many isomers (including geometrical and optical) are possible for bromochlorocyclobutane?
- **4.** How many stereoisomers are possible for dichlorocyclobutane?
- 5. How many enantiomeric pairs are possible in bromochlorocyclopentane?
- 6. How many alkyl bromides would yield isopentane on reaction with Grignard reagent followed by treatment with water?
- 7. How many free radicals can be produced during following reaction (ignoring resonating structure)?

8.
$$D \longrightarrow \frac{Br_2}{brine} \rightarrow (B)$$

Find the number of fractions obtained after fractional distillation of product mixture.

9.
$$\begin{array}{c}
 & H_2/Ni \\
\hline
 & (a) \\
\hline
 & O_3 \\
\hline
 & (b)
\end{array}$$
(Cl₂/hv Monochlorination (All isomers)

Calculate sum of number of products formed in the reaction a, b and c.

- **10.** How many alkenes, alkynes, alkadienes can be hydrogenated to form isopentane (include all isomers)?
- 11. Find the total number of cyclic isomers possible for a hydrocarbon with the molecular formula C₄H₆.
- 12. Write total number of hydrogen atoms on all the carbon atoms which are connected directly by a single bond to benzylic carbon (carbon connected to benzene ring) in the product

13. Number of monochloro derivatives (excluding stereoisomers), dichloro derivatives and trichloro derivatives of cyclopentane are n_1 , n_2 and n_3 . Find the value of $(n_1 + n_2)/n_3$.

Find the number of alkenes produced in the reaction given.

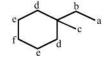
15. Find the total no. of alkynes that on catalytic reduction gives 3-ethyl-4-methylheptane





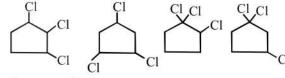
SOLUTIONS

1. (6)



There are six positions in the given compound which will give monochloroderivatives.

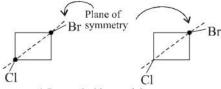
2. (4)



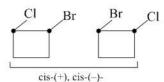
3. (7) Seven possible isomers are

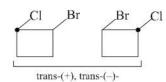


1-Bromo-1-chlorocyclobutane (Stereoisomerism not possible)



1-Bromo-3-chlorocyclobutane (Both are optically inactive)

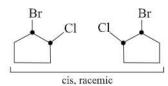


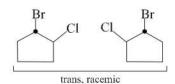


1-Bromo-2-chlorocyclobutane

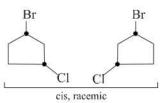
4. (5) 1, 3-Dichlorocyclobutane can exist in *cis* and *trans* forms. *trans*-1, 2-Dichlorocyclobutane can exist in (+) – and (–) – forms. However, *cis*-1, 2 Dichlorocyclobutane has a plane of symmetry and hence it can exist as *meso* isomer.

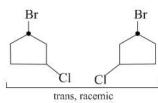
5. (4)





1-Bromo-2-cyclopentane





1-Bromo-3-cvclopentane

6. (4) All alkyl bromides having carbon skeleton of isopentane (2-methylbutane (CH₃)₂CHCH₂CH₃) will give isopentane via Grignard reagent.

$$\begin{array}{ccc} \operatorname{CH}_3 & \operatorname{CH}_3 \\ | & | \\ \operatorname{BrCH}_2\operatorname{CHCH}_2\operatorname{CH}_3 & \operatorname{CH}_3\operatorname{CBrCH}_2\operatorname{CH}_3 \end{array}$$

8. (6) D
$$\stackrel{\text{Br}_2}{=}$$
 aq.NaCl

9. (6)
$$\begin{array}{c} H_2/hv \\ \hline \end{array}$$
 (1)
$$\begin{array}{c} Cl_2/hv \\ \hline \end{array}$$

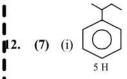
$$\begin{array}{c|c} \text{(b)} & O_3 & & \text{CH}_2 - \text{CH}_2\text{Cl} \\ \text{CHO} & & & & \\ \text{OHC} & -\text{C} - \text{CHO} + \text{CH}_2\text{O} \\ \text{CHO} & & & \\ \end{array}$$

$$a = 1$$
; $b = 2$; $c = 3$ Sum $= 1 + 2 + 3 = 6$

10. (6)
$$C = C - C - C$$
, $C - C = C - C$

 (5) The number of cyclic isomers for a hydrocarbon with molecular formula C₄H₆ is 5.
 The structures are

$$\hfill \Box$$
 , $\hfill \Box$, $\hfill \Box$





2

43. (1)

and

$$\therefore \frac{n_1 + n_2}{n_3} = \frac{1+3}{4} = \frac{4}{4} = 1$$

14. (5)

[Five]

3-Ethyl-4-methylheptane

(i)
$$CH = C - CH_2 - CH - CH - CH_2 - CH_3$$

 $CH_3 CH_2CH_3$

(ii)
$$CH_3 - CH_2 - CH_2 - CH - CH - CH_2CH_3$$

 CH_3 $C = CH$

Alkynes (i), (ii) and (iii) produces 3-ethyl-4-methyl heptane on catalytic reduction.

